MCDB 3145 Molecular & Cell Biology II

Part I. Cell Components and Compartments

Spring 2016, Stowell





Animal cell (eukaryote)









Table 1.1 A Comparison of Prokaryotic and Eukaryotic Cells

Features held in common by the two types of cells:

- Plasma membrane of similar construction
- Genetic information encoded in DNA using identical genetic code
- Similar mechanisms for transcription and translation of genetic information, including similar ribosomes
- Shared metabolic pathways (e.g., glycolysis and TCA cycle)
- Similar apparatus for conservation of chemical energy as ATP (located in the plasma membrane of prokaryotes and the mitochondrial membrane of eukaryotes)
- Similar mechanism of photosynthesis (between cyanobacteria and green plants)
- Similar mechanism for synthesizing and inserting membrane proteins
- Proteasomes (protein digesting structures) of similar construction (between archaebacteria and eukaryotes)

Features of eukaryotic cells not found in prokaryotes:

- Division of cells into nucleus and cytoplasm, separated by a nuclear envelope containing complex pore structures
- Complex chromosomes composed of DNA and associated proteins that are capable of compacting into mitotic structures
- Complex membranous cytoplasmic organelles (includes endoplasmic reticulum, Golgi complex, lysosomes, endosomes, peroxisomes, and glyoxisomes)
- Specialized cytoplasmic organelles for aerobic respiration (mitochondria) and photosynthesis (chloroplasts)
- Complex cytoskeletal system (including microfilaments, intermediate filaments, and microtubules) and associated motor proteins
- Complex flagella and cilia
- Ability to ingest particulate material by enclosure within plasma membrane vesicles (phagocytosis)
- Cellulose-containing cell walls (in plants)
- Cell division using a microtubule-containing mitotic spindle that separates chromosomes
- Presence of two copies of genes per cell (diploidy), one from each parent
- Presence of three different RNA synthesizing enzymes (RNA polymerases)
- Sexual reproduction requiring meiosis and fertilization



Basic components of a eukaryotic cell

Model organisms commonly used in cell biology research









100







2

Many cell types in the human body



Pluripotent stem cells can be derived from embryos



Differentiation of embryonic stem (ES) cells into various cell types



Pluripotent stem cells can be derived from adults (induced pluripotent stem cells, iPS cells)



Differentiation of human induced pluripotent stem (iPS) cells





<u>Shen lab</u>

Differentiated fat cells

Differentiated neurons

The Nobel Prize in Physiology or Medicine 2012



Photo: U. Montan Sir John B. Gurdon

Photo: U. Montan Shinya

The Nobel Prize in Physiology or Medicine 2012 was awarded jointly to Sir John B. Gurdon and Shinya Yamanaka "for the discovery that mature cells can be reprogrammed to become pluripotent"

Yamanaka



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Carbohydrates/ glycans

Simple sugars/ monosaccharides



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Oligosaccharides

Sucrose



Lactose



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Polysaccharides





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Triacylglycerol (triglyceride)



Long-term energy storage in animals

Lipids



Phospholipid





Proteins/polypeptides



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Properties of side chains (R groups):

Hydrophilic side chains act as acids or bases which tend to be fully charged (+ or –) under physiologic conditions. Side chains form ionic bonds and are often involved in chemical reactions.



Properties of side chains:

Hydrophilic side chains tend to have partial + or – charge allowing them to participate in chemical reactions, form H-bonds, and associate with water.

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Potential phosphorylation sites



Hydrophobic side chain consists almost entirely of C and H atoms. These amino acids tend to form the inner core of soluble proteins, buried away from the aqueous medium. They play an important role in membranes by associating with the lipid bilayer.

Side chains with unique properties

(Gly or G) Side chain consists only of hydrogen atom and can fit into either a hydrophilic or hydrophobic environment.Glycine often resides at sites where two polypeptides come into close contact.

Though side chain has polar, uncharged character, it has the unique property of forming a covalent bond with another cysteine to form a disulfide link.

Cysteine

(Cys or C)

 $\begin{array}{c}
\mathsf{CH}_2 - \mathsf{CH}_2 \\
\mathsf{CH}_2 & \mathsf{CH} - \mathsf{C} - \mathsf{O}^- \\
\mathsf{N} & \mathsf{U} \\
\mathsf{N} & \mathsf{O} \\
^+ \mathsf{H}_2 \\
\mathsf{Proline} \\
\mathsf{(Pro or P)}
\end{array}$

Though side chain has hydrophobic character, it has the unique property of creating kinks in polypeptide chains and disrupting ordered secondary structure.

Cysteine



Clicker questions will be uploaded into D2L weekly.

A practice exam will be provided in D2L one week before each actual exam.

TA Office Hours

TA Name	Time	Location
Kati Buchtel	Tuesday 4 - 5:30	Gold A1B16 (Cell Bio Lab Room)
Christine Crotzer	Tuesday 5:30 - 6:30	Gold A1B16 (Cell Bio Lab Room)
Deanna Langager	Thursday 10 - 11:00	2nd Floor Interaction Room

Joy Power's Office is B126D in Porter. Stop by anytime the door is open. If you need to set an appointment, use email: <u>Joy.Power@colorado.edu</u>.

Protein folding – to find the lowest energy state (native conformation)



Noncovalent bonds



Secondary structures - alpha helix

Part (a) from J. Banaver, Ann Rev. Biophys. Biomol. Struct. 36:268, 2007, Figs. 4b Figure created by Timothy Lezon. Reprinted with permission of Annual Reviews.



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Secondary structures – beta sheet



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Tertiary structure of a protein



Quaternary structure of a protein complex



Hemoglobin

Protein domains



From Liisa Holm and Chris Sander, Structure 5:167, (C) 2007, with permission from Elsevier.
How do proteins fold?



Christian Anfinsen 1972 Nobel Prize in Chemistry

Denatured RNase spontaneously folds into its native conformation



Protein folding – to find the lowest energy state (native conformation)



Chaperones assist protein folding in the cell



DnaK/Hsp70 chaperones promote protein folding



The chaperonin system



Heat shock response



Prion protein (PrP)





PrP^C

Alzheimer's Disease



homas Deerinck, NCMIR/Photo Researchers, Inc.

Amyloid plaque

Neurofibrillary tangle (NFT)

Normal

Alzheimer's





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The plasma membrane



Courtesy J. D. Robertson

Membrane - lipid bilayer (Formed by mixing phospholipids and water)

Phospholipids: building blocks of membranes



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Phospholipids: building blocks of membranes





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Lipid asymmetry



Table 4.1 Lipid Compositions of Some Biological Membranes*

Lipid	Human erythrocyte	Human myelin	Beef heart mitochondria	E. coli
Phosphatidic acid	1.5	0.5	0	0
Phosphatidylcholine	19	10	39	0
Phosphatidyl-				
ethanolamine	18	20	27	65
Phosphatidylgycerol	0	0	0	18
Phosphatidylserine	8.5	8.5	0.5	0
Cardiolipin	0	0	22.5	12
Sphingomyelin	17.5	8.5	0	0
Glycolipids	10	26	0	0
Cholesterol	25	26	3	0

*The values given are weight percent of total lipid. Source: C. Tanford, The Hydrophobic Effect, p. 109, copyright 1980, John Wiley & Sons, Inc. Reprinted by permission of John Wiley & Sons, Inc.

Dynamic plasma membrane



Courtesy Susan Jo Burwen

Fluid-mosaic model



Synaptic vesicle model



Fluid-mosaic model



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Membrane proteins

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Protein-membrane interactions



From Carola Hunte and Sebastian Richers, Curr. Opin. Struct. Biol. 18: 407, 2008, © 2008, with permission of Elsevier Science

Structure of an integral membrane protein



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Orientation of residues in the membrane



From Anna C.V. Johansson and Erik Lindahl, Biophys. J. 91:4459, 4453, © 2006, with permission from Elsevier.





Lipid-anchored proteins



Doc2b: a calcium-regulated peripheral membrane protein



Solubilization of membrane proteins






Temperature and membrane structure



(a)

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Fluid



Crystalline gel

Table 4.2 Melting Points of the Common 18-Carbon Fatty Acids

Fatty acid	cis Double bonds	M.p.(°C)
Stearic acid	0	70
Oleic acid	1	13
Linoleic acid	2	-9
Linolenic acid	3	-17
Eicosapentanoic acid (EPA)*	5	-54

*EPA has 20 carbons.

Lipid rafts



From D.E. Saslowsky, et al, J. Biol. Chem. 277, 26966.26970, July 26, © 2002 The American Society for Biochemistry and Molecular Biology.





Membrane protein dynamics





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Membrane protein mobility



Polarized cell





Membrane proteins

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Molecular transport across membranes





Molecular transport across membranes





(a) Hypotonic solution

Net water loss Cell shrinks



No net loss or gain



(b) Hypertonic solution

(c) Isotonic solution

Aquaporin – water channel



Aquaporin – water channel



From Benoit Roux and Klaus Schulten, Structure 12:1344, © 2004, by permission of Elsevier.

Molecular transport across membranes



Facilitative glucose transporter



Facilitative glucose transporter



Transporters can work in both directions



Active glucose transport (Na⁺/glucose cotransporter)



Active glucose transport



Molecular transport across membranes



Ion channel



Ion transport is electrogenic







From T.D. Lamb, H.R. Matthews and V. Torre, J. Physiology 372:319, © 1986, reproduced with permission from John Wiley & Sons.

K⁺ ion channel





Voltage-gated K⁺ channel





Reprinted with permission from Stephen B. Long, et al., Science 309: 867, 899, 2005, courtesy of Roderick MacKinnon; © 2005, reprinted with permission from AAAS.



Voltage-gated K⁺ channel



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"closed"

Table 4.3 Ion Concentrations Inside and Outside of a Typical Mammalian Cell

	Extracellular concentration	Intracellular concentration	lonic gradient
Na ⁺	150 mM	10 mM	15×
K +	5 mM	140 mM	28 ×
Cl⁻	120 mM	10 mM	12×
Ca ²⁺	10 ⁻³ M	10 ⁻⁷ M	10,000×
H+	10 ^{−7.4} M	10 ^{-7.2} M	Nearly $2 \times$
	(pH of 7.4)	(pH of 7.2)	

The ion concentrations for the squid axon are given on page 177.

Na⁺/K⁺ pump



Na⁺/K⁺ pump


Na⁺/K⁺ pump



Cystic fibrosis transmembrane conductance regulator (CFTR)



Airway epithelium from normal individual

Airway epithelium from CF patient





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Synaptic transmission







Don W. Fawcett / T. Reese/Photo Researchers, Inc.



- Membrane of postsynaptic target cell
- Synaptic vesicles
- Synaptic cleft
- Terminal knob of presynaptic neuron



Synaptic transmission



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Shen lab

Cytoplasmic organelles



Plant root cell



~50% of the cell volume is in membrane-bound organelles



From Alain Rambourg and Yves Clermont, Eur. J. Cell Biol. 51:196, 1990



Why does the cell need such diverse organelles?