Reading Material Lecture 1



CLICK TO OPEN PDF FILE

Article 1:

http://dosequis.colorado.edu/Courses/MicroNano/articles/Lecture1_Article1.pdf

Article 2:

http://dosequis.colorado.edu/Courses/MicroNano/articles/Lecture1_Article2.pdf

Article 3:

http://dosequis.colorado.edu/Courses/MicroNano/articles/Lecture1_Article3.pdf

Article 4:

http://dosequis.colorado.edu/Courses/MicroNano/articles/Lecture1_Article4.pdf

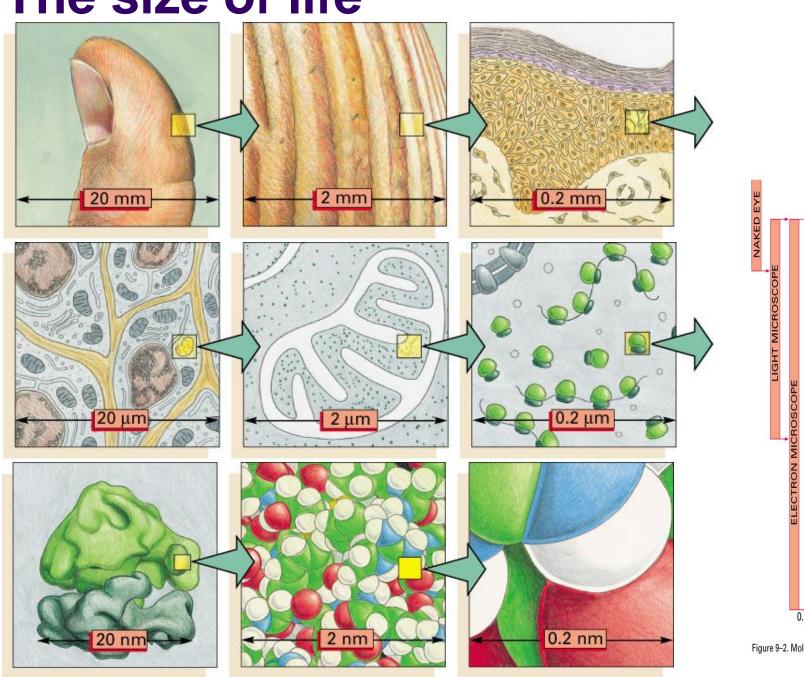
Moc/Bio and Nano/Micro Lee and Stowell

Moc/Bio-Lecture 1

Biological Molecules Nucleic Acids (nucleotides) Proteins (amino acids) Carbohydrates (sugars) Lipids (fatty acids) Chemistry of Biomolecules Peptides Phosphates Esters etc



The size of life



100 nm Virus ribosome 10 nm globular protein 1 nm 0.1 nm (1 Å) Figure 9–2. Molecular Biology of the Cell, 4th Edition.

^{1 cm}≣

1 mm

100 µm

10 µm

plant cell

animal cell



The chemistry of life

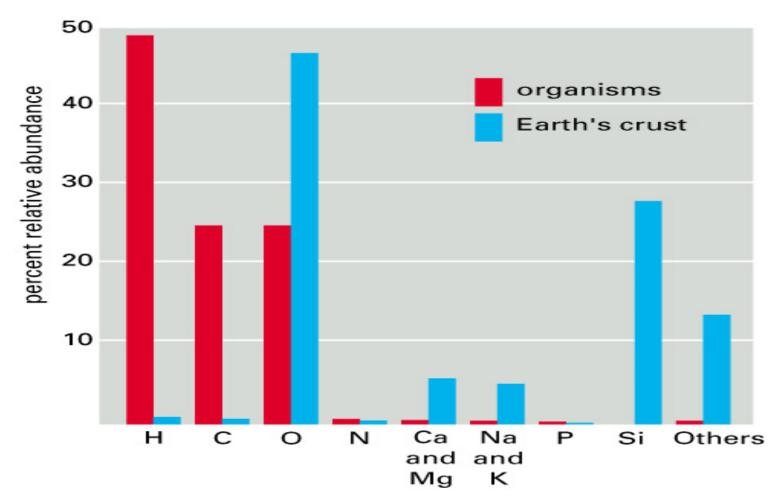


Figure 2–3. Molecular Biology of the Cell, 4th Edition.

Something about 4

- There are four fundamental elements in biology
 - Carbon
 - Nitrogen
 - Oxygen
 - Hydrogen
- There are four fundamental molecules in biology
 - Nucleic acids
 - Amino acids
 - Carbohydrates
 - Fatty acids
- The flow of information and function in the cell is
 - From DNA
 - To RNA
 - To proteins
 - To function



Chemical Bonds



- Covalent Bonds
 - C-C
- Ionic Bonds
 - Na⁺Cl⁻
- Hydrogen Bonds
 - water
- Van der Waals
 - oils

Nucleic acids

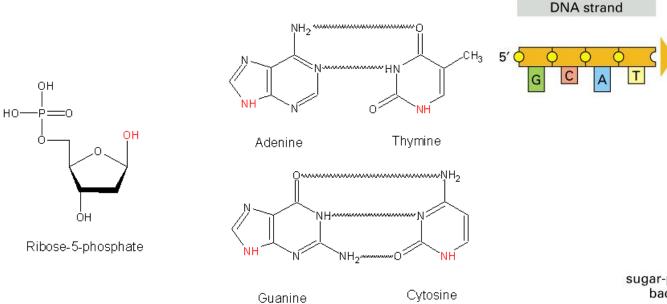
- Deoxyribonucleic acids (DNA)
 - Structure
 - Function
- Ribonucleic acids (RNA)
 - Structure
 - Function



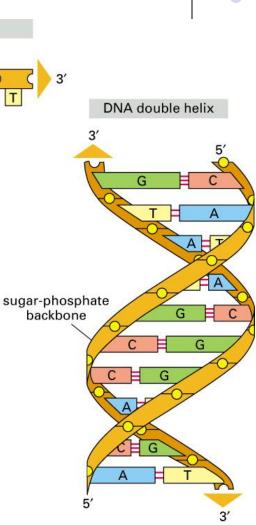
DNA-structure

• Deoxyribonucleic acids





- Double helix (Watson, Crick, Wilkins and Franklin)
 - A=T, G=C Chargaff



Chemical recognition

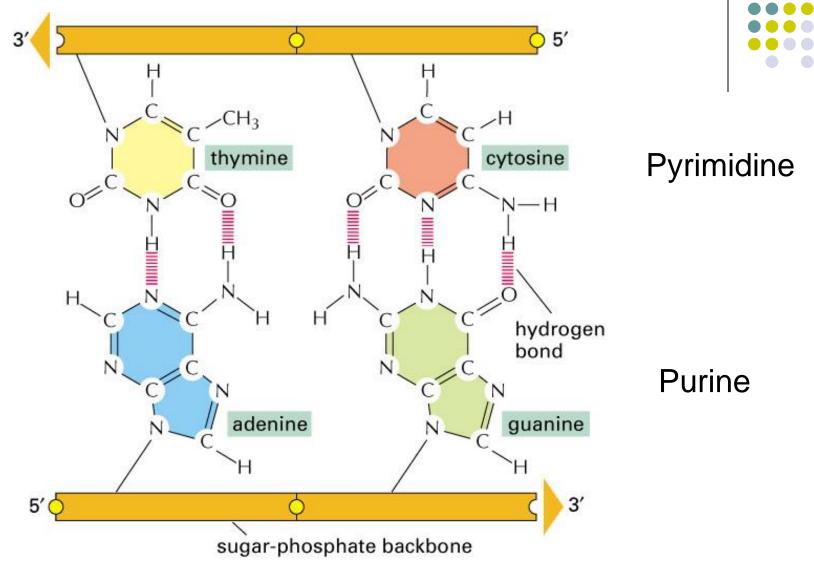


Figure 4–4. Molecular Biology of the Cell, 4th Edition.

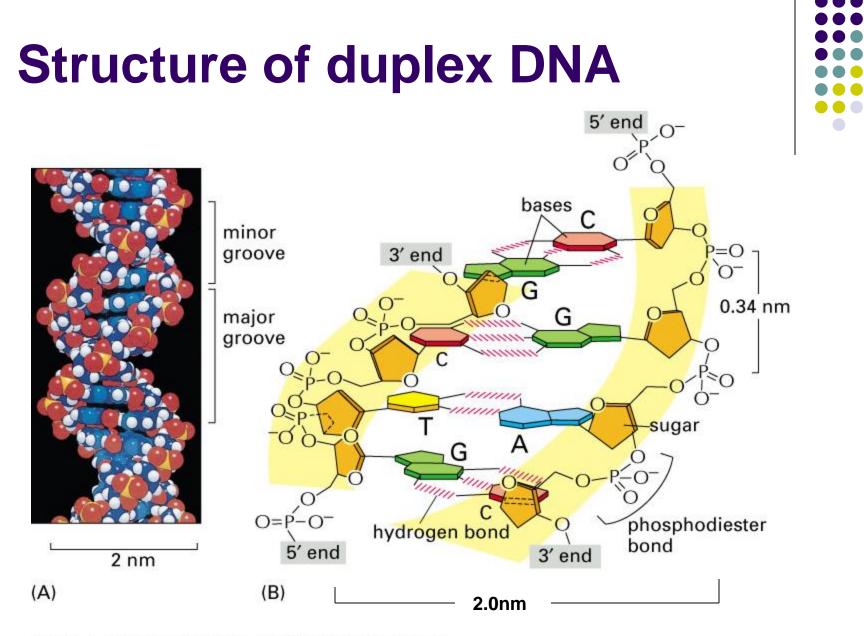


Figure 4–5. Molecular Biology of the Cell, 4th Edition.

Major Minor differences



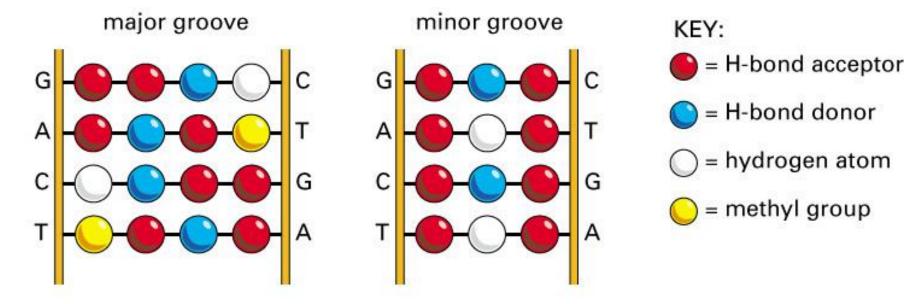
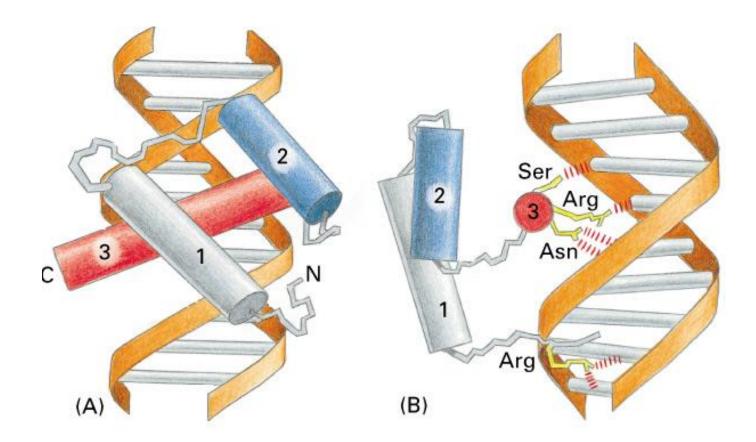


Figure 7–8. Molecular Biology of the Cell, 4th Edition.

Recognition







DNA-biology

- Storage and replication of genetic information
 - Coding and noncoding strand
 - High fidelity replication
- Encodes for proteins and the regulation of protein expression
- Encodes for siRNA's





Storage of information

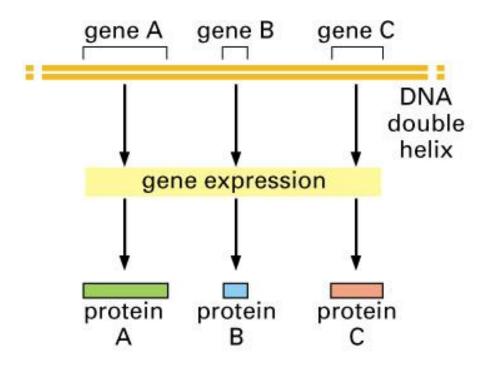


Figure 4-6. Molecular Biology of the Cell, 4th Edition.



High fidelity replication

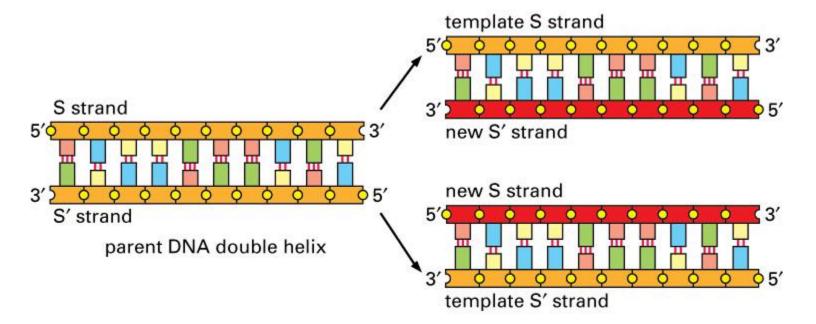


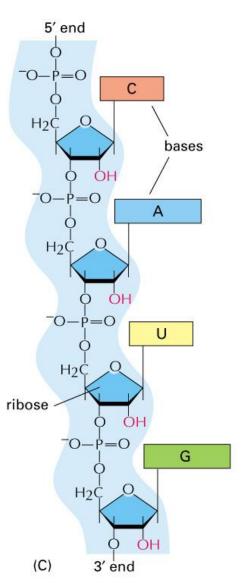
Figure 4–8. Molecular Biology of the Cell, 4th Edition.

Only 1 in 10⁹ ! errors

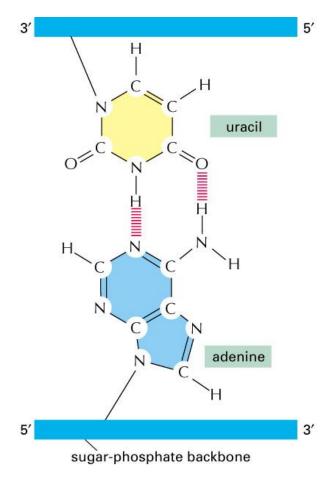


RNA-structure

- Ribonucleic acid
 он
 но-р́то
 он
 <
- Also complements sequence (G-C A-U)
- Fold into more complex 3D structures
 - tRNA's, ribozymes, regulatory loops



RNA bases





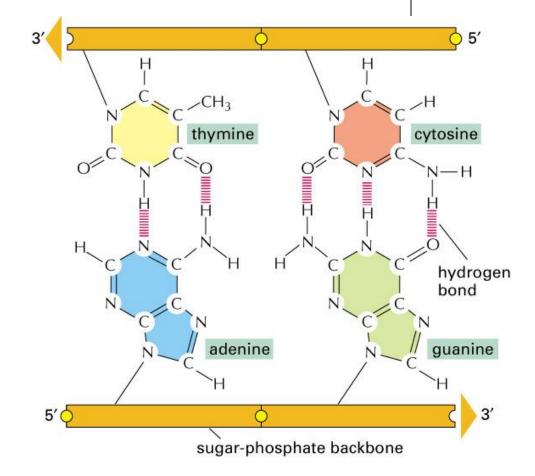


Figure 4-4. Molecular Biology of the Cell, 4th Edition.

RNA structures



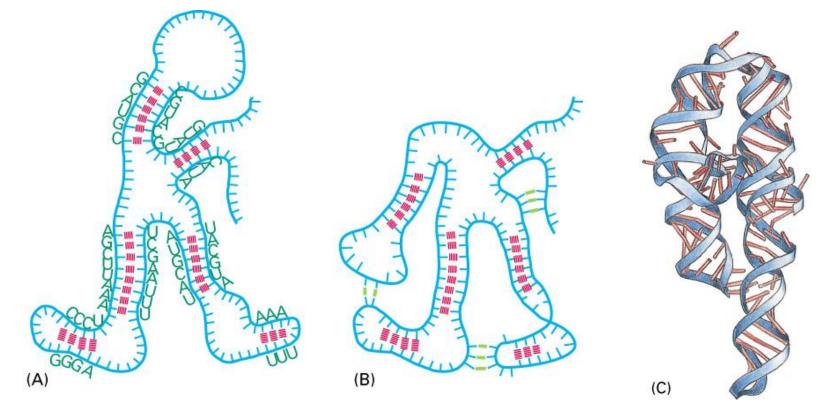
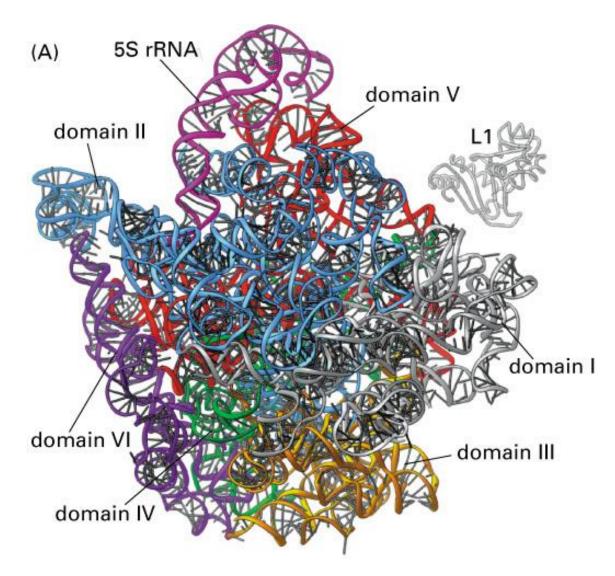
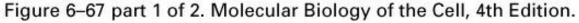


Figure 6-6 part 1 of 2. Molecular Biology of the Cell, 4th Edition.

Figure 6-6 part 2 of 2. Molecular Biology of t

The ultimate ribozyme







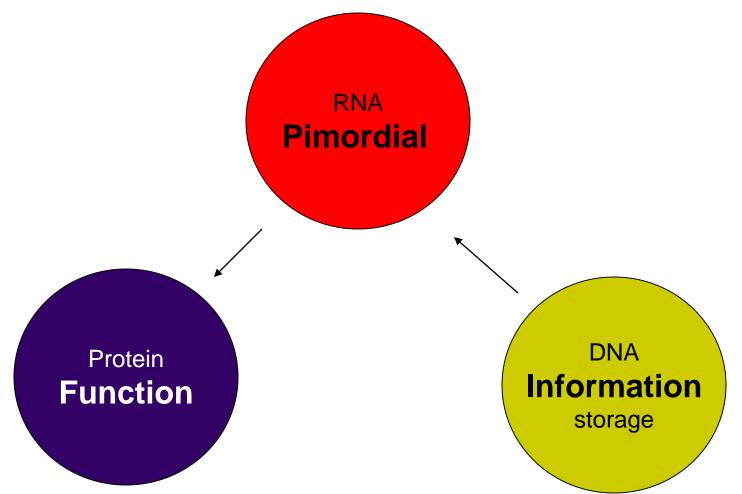
RNA-biology

- Primordial life
 - The first enzymes were ribozymes
- Intermediate for protein expression
- Catalyzes peptide synthesis
- Second level of protein regulation





Tripartite division of labor



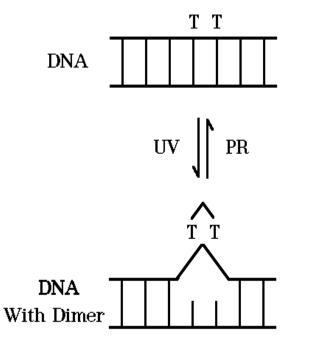
Reactions of Nucleotides

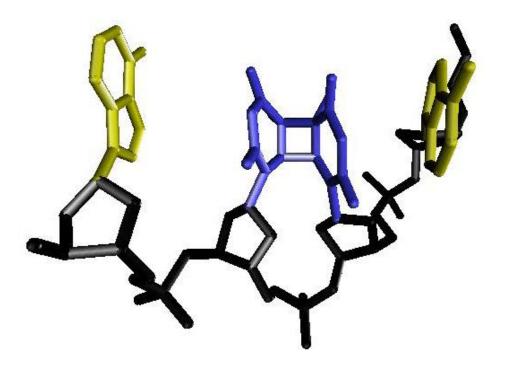
- Chemical hydrolysis
 - RNA phosphoryl hydrolysis greater than DNA
 - Cleaves both the phosphoryl and the N-glycosidic
- Oxidative damage
 - Most prevalent
 - Deoxyribose cleavage and base loss
 - Heterocycle cleavage
- Photo induced dimer formation
 - T-T formation from T T in sequence
 - Repair enzyme photolyase



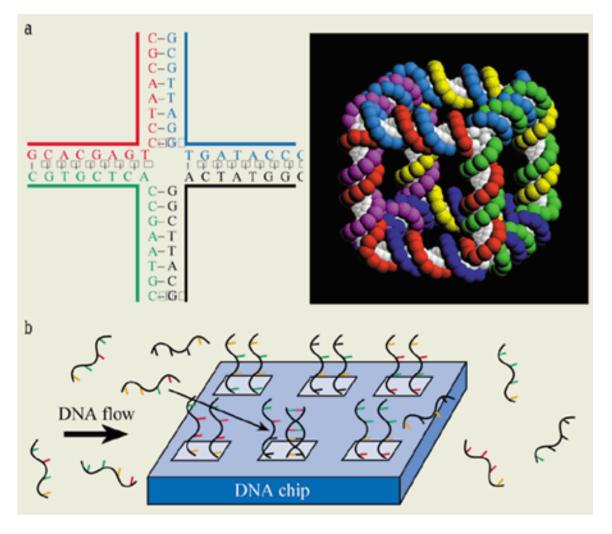
Thymidine dimer and photolyase

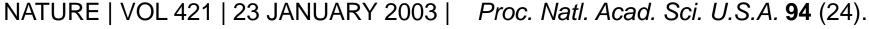






Examples: Exploiting base pair specificity









L-Amino Acids-protein

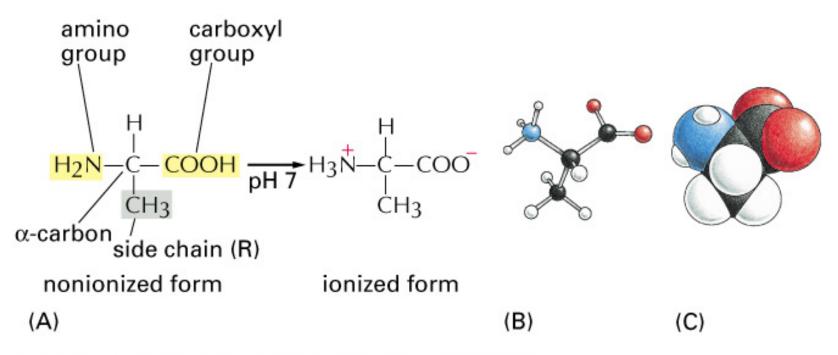


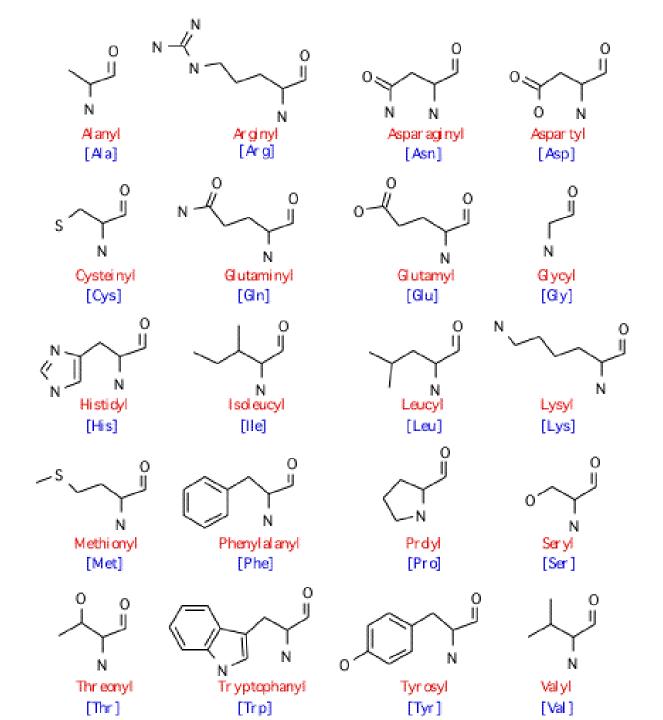
Figure 2–23. Molecular Biology of the Cell, 4th Edition.

The natural amino acids

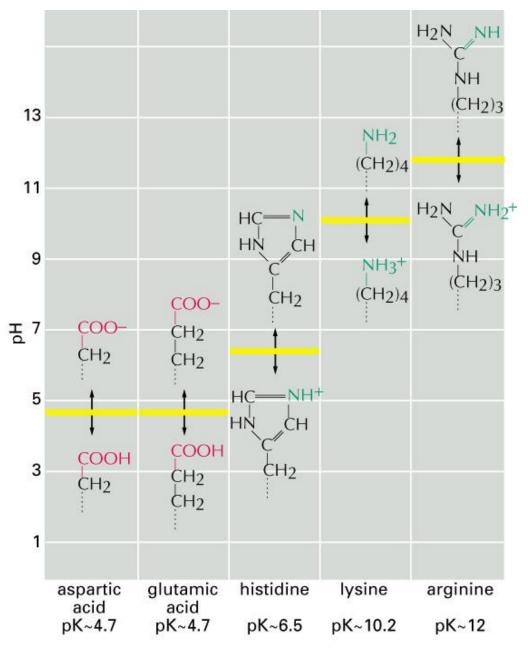
AMINO ACID			SIDE CHAIN	AMINO ACID		SIDE CHAIN	
Aspartic acid	Asp	D	negative	Alanine	Ala	А	nonpolar
Glutamic acid	Glu	Ε	negative	Glycine	Gly	G	nonpolar
Arginine	Arg	R	positive	Valine	Val	V	nonpolar
Lysine	Lys	Κ	positive	Leucine	Leu	L	nonpolar
Histidine	His	Н	positive	Isoleucine	lle	1	nonpolar
Asparagine	Asn	Ν	uncharged polar	Proline	Pro	Ρ	nonpolar
Glutamine	Gln	Q	uncharged polar	Phenylalanine	Phe	F	nonpolar
Serine	Ser	S	uncharged polar	Methionine	Met	М	nonpolar
Threonine	Thr	Т	uncharged polar	Tryptophan	Trp	W	nonpolar
Tyrosine	Tyr	Υ	uncharged polar	Cysteine	Cys	С	nonpolar
POLAR AMINO ACIDS				NONPOLAR AMINO ACIDS			

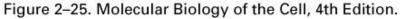
Figure 3–3. Molecular Biology of the Cell, 4th Edition.







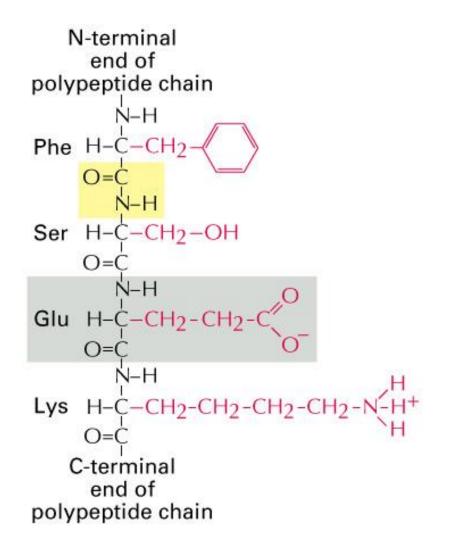






Variance

Polypetides (proteins)







Proteins-structure

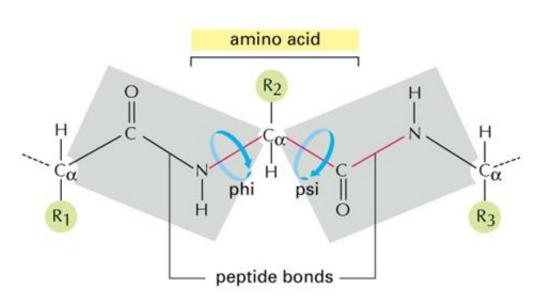
- Primary structure
 - Peptide bond
- Secondary structure
 - H-bond
 - Helix, sheet, extended
- Tertirary structure
 - Globular folds
 - VDW-forces

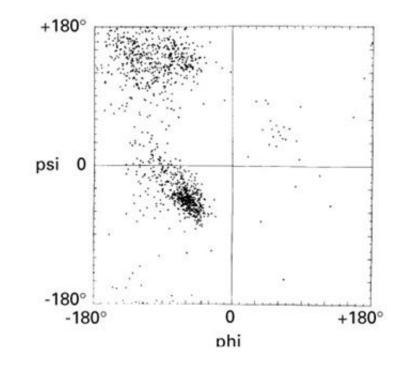


The peptide bond



Resonance





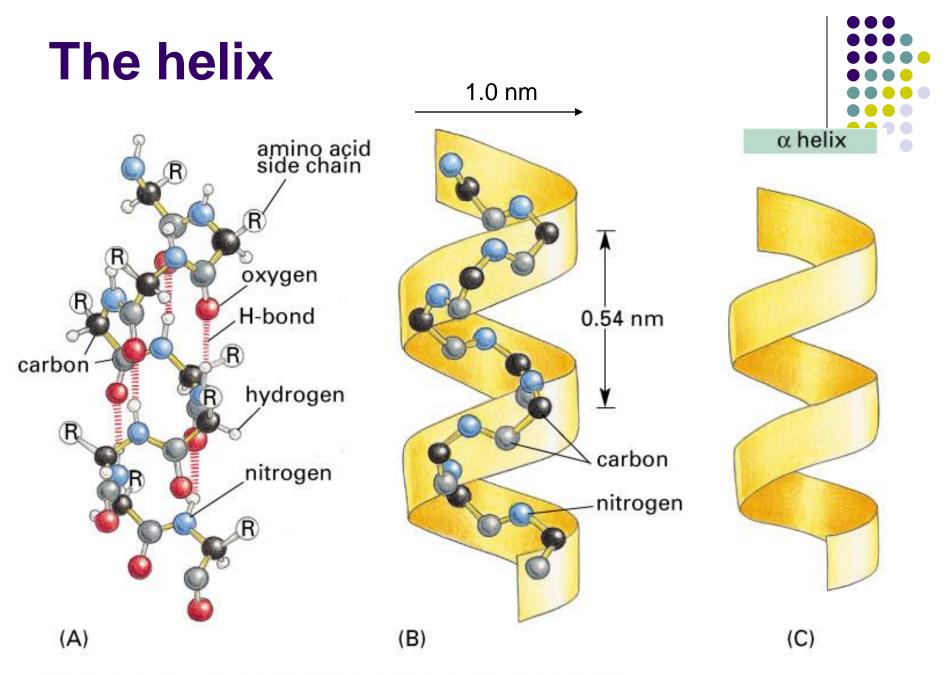


Figure 3–9 part 1 of 2. Molecular Biology of the Cell, 4th Edition.

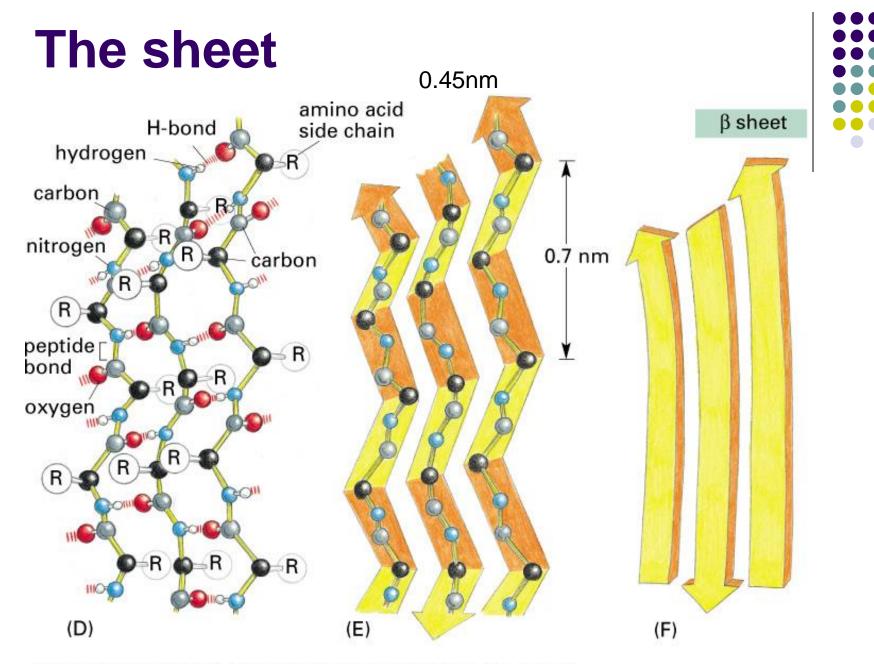


Figure 3–9 part 2 of 2. Molecular Biology of the Cell, 4th Edition.

Add them together and....

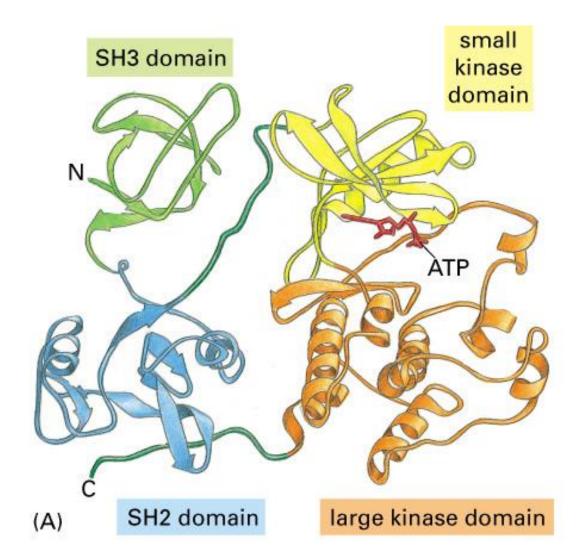




Figure 3–12 part 1 of 2. Molecular Biology of the Cell, 4th Edition.

Proteins-biology

- The workhorses of the cell
 - Catalysts
 - Energy production
 - Regulation
 - Assembly
 - Motility
 - Repair
 - Etc, etc.....



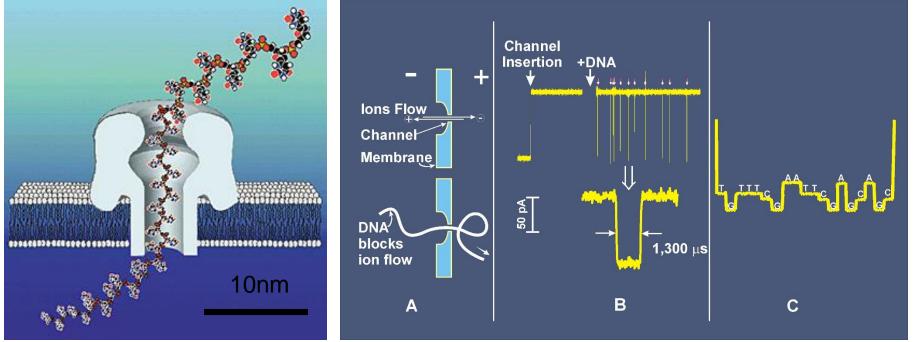
Reactions of proteins

- Sulfur oxidation (Cys disulfides, S-thiolation; Met sulfoxide)
- Protein carbonyls (side chain aldehydes, ketones)
- Tyrosine crosslinks, chlorination, nitrosation, hydroxylation
- Tryptophane oxidation
- Hydro(pero)xy derivatives of aliphatic amino acids
- Chloramines, deamination
- Amino acid interconversions (*e.g.*, His to Asn; Pro to OH-Pro)
- Amino acid oxidation adducts (*e.g.*, *p*-hydroxyphenylacetaldehyde)
- Glycoxidation adducts (*e.g.*, carboxymethyllysine)
- Cross-links, aggregation, peptide bond cleavage





Nano-DNA-sequencer



Proc Natl Acad Sci USA 93 (24)

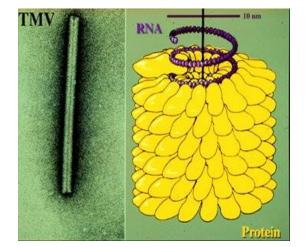
DNA encodes for Proteins

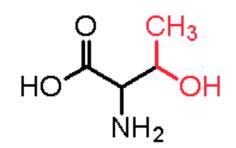


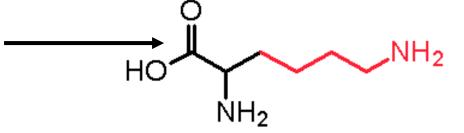
- 3 bases make a codon
- Each codon encodes a single amino acid
- Change the codon \rightarrow change the amino acid

Example: Alter the chemistry of a known biomolecule.

- The three potential external labeling positions
 - N-terminus
 - C-terminus
 - 63-66 loop
- Conversion of Threonine to Lysine Change DNA-Codon from ACC to AAG

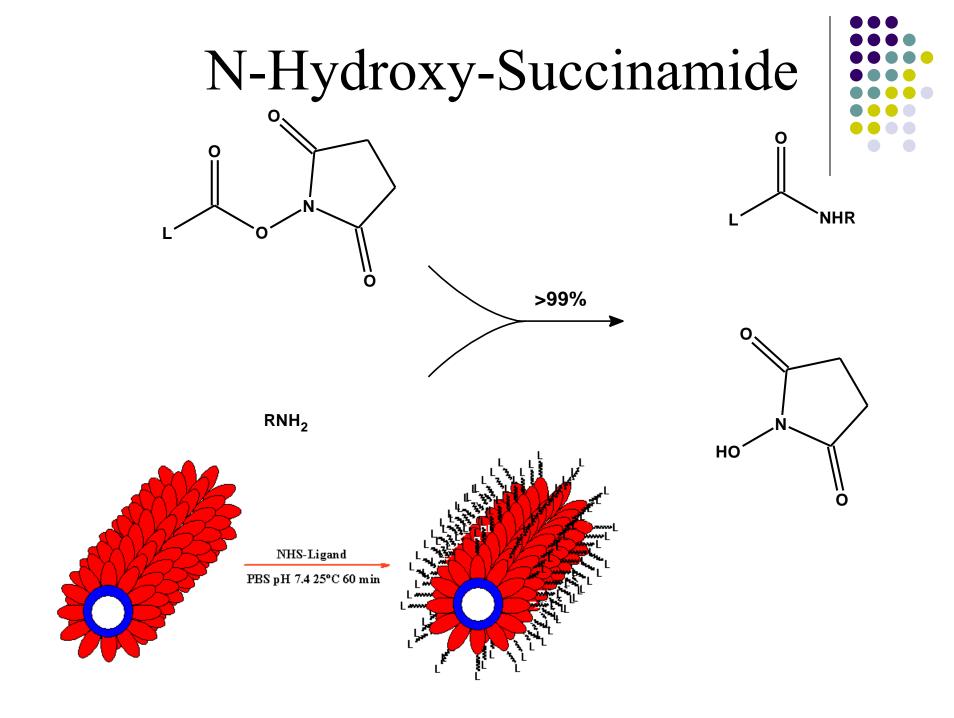




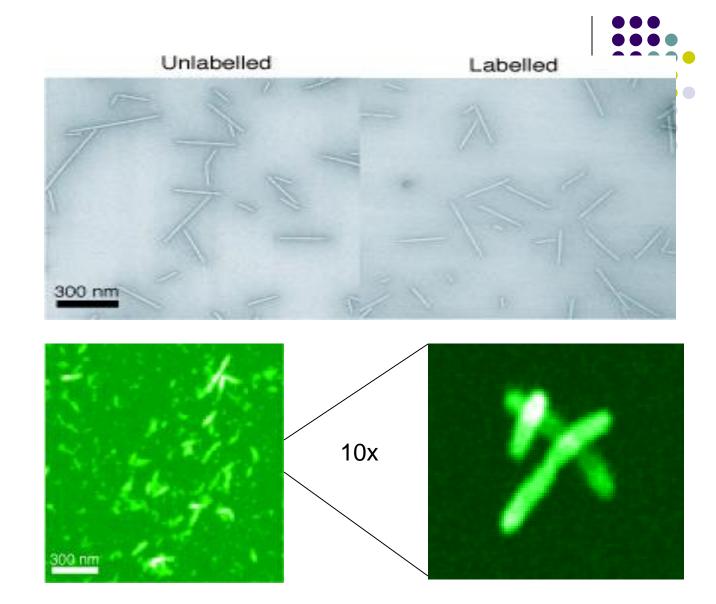


Nanotechnology 13 (2002) 541-544



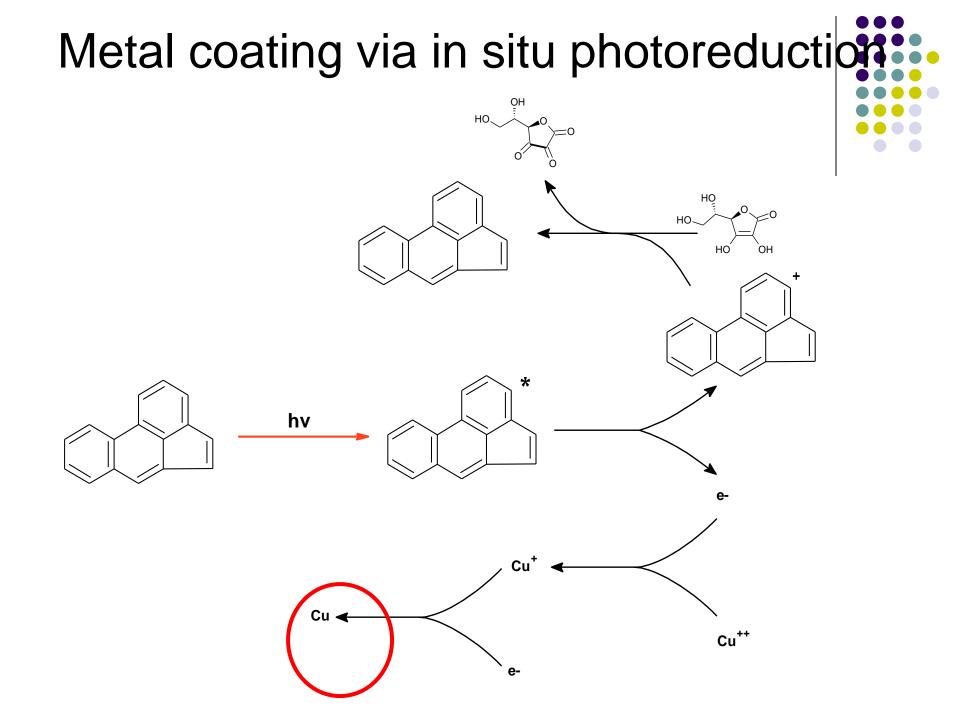






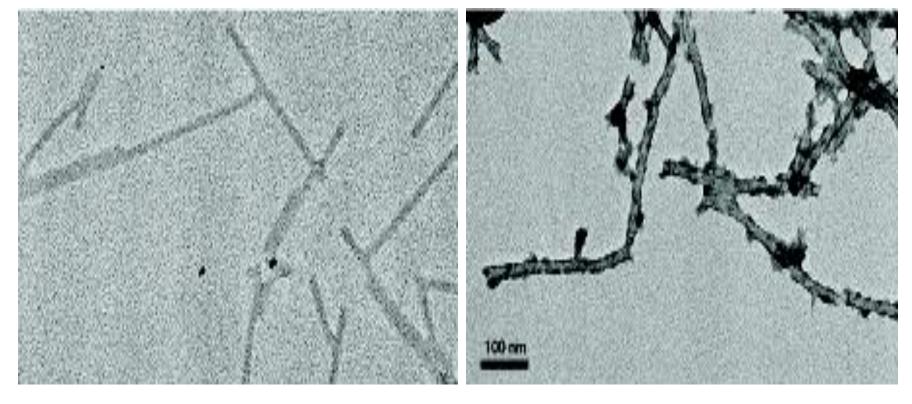
Fluorescence Confocal

Fluorescent labeling

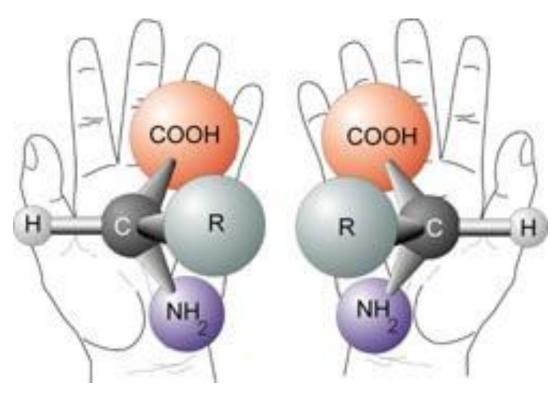


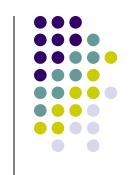


Metal coated TMV -hv +hv



Workshop Question?





Why L-amino acids? Proteins from living organism consist of exclusively L-amino acids. If I synthesized a D-amino acid version of a known enzyme what might you predict about this enzyme.

Carbos structure

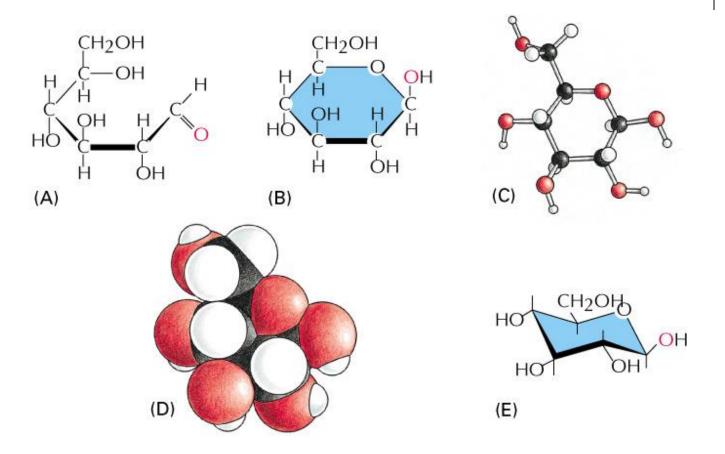


Figure 2–18. Molecular Biology of the Cell, 4th Edition.



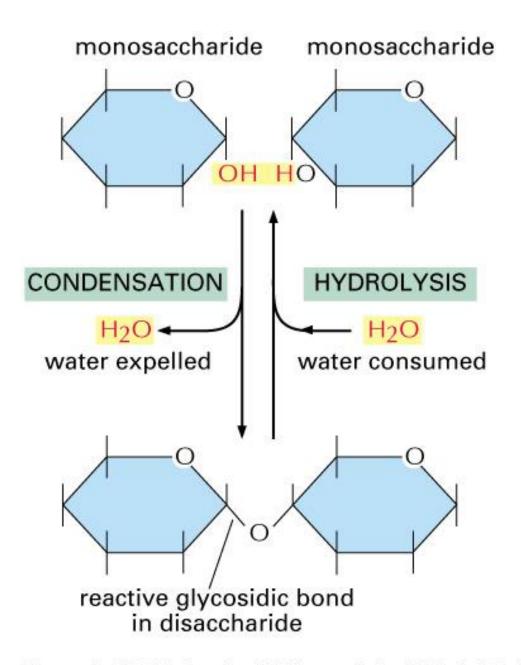




Figure 2–19. Molecular Biology of the Cell, 4th Edition.

Reactions of carbos

- Primary oxidation as in Nucleotides
- Hydrolysis to shorter chain polys





Carbohydrates-biology

- Polysaccharides
 - Protein modification
 - Cell surface modification
 - The food stuff of the cell

Protein modification



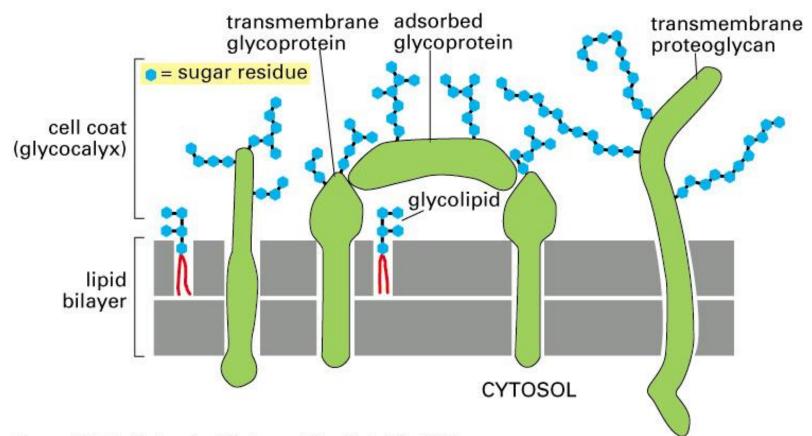
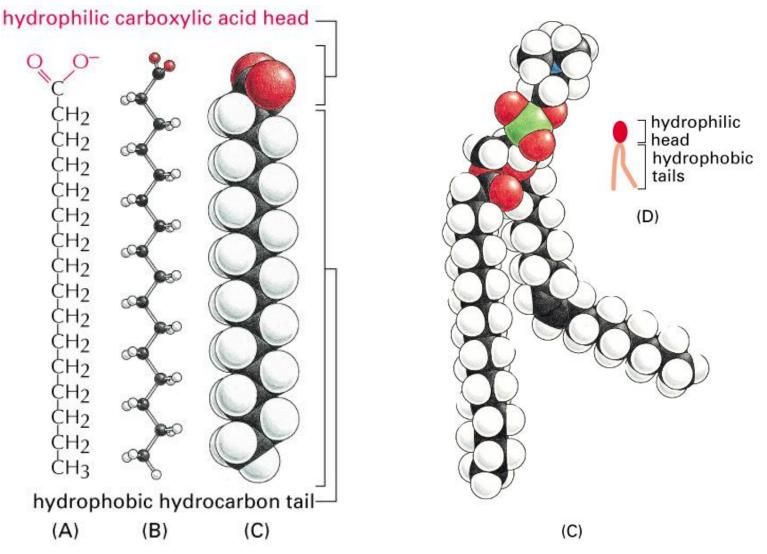


Figure 10–45. Molecular Biology of the Cell, 4th Edition.

Fatty acids







The cell membrane



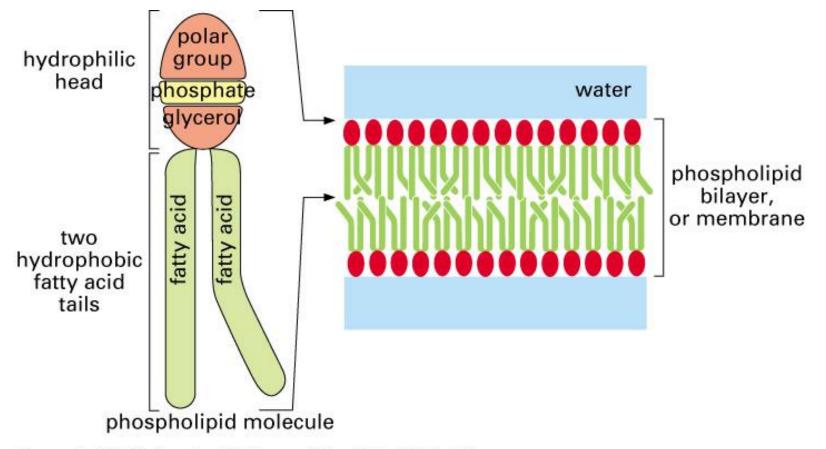


Figure 2–22. Molecular Biology of the Cell, 4th Edition.

Lipid variation-head to tail

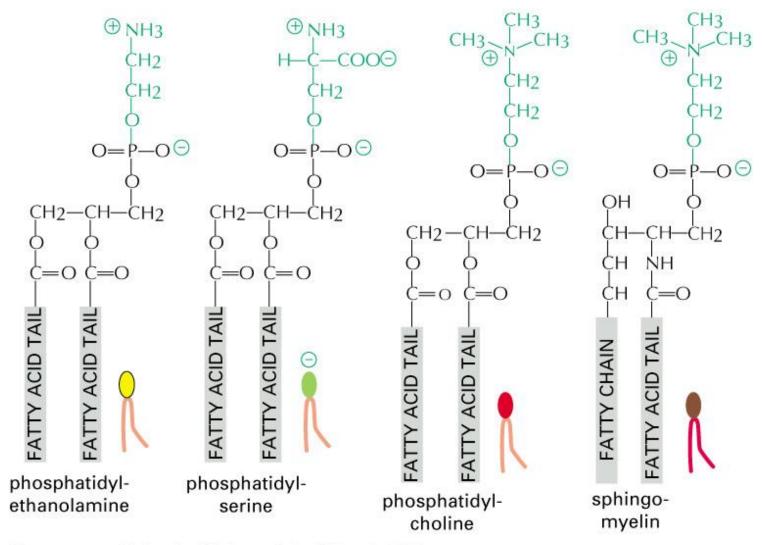


Figure 10–12. Molecular Biology of the Cell, 4th Edition.



Fatty-acids biology



- The cell membrane (barrier to the outside)
 - Regulates uptake of nutrients and waste
- Regulate membrane proteins
 - Modification
 - Stabilization

The membrane



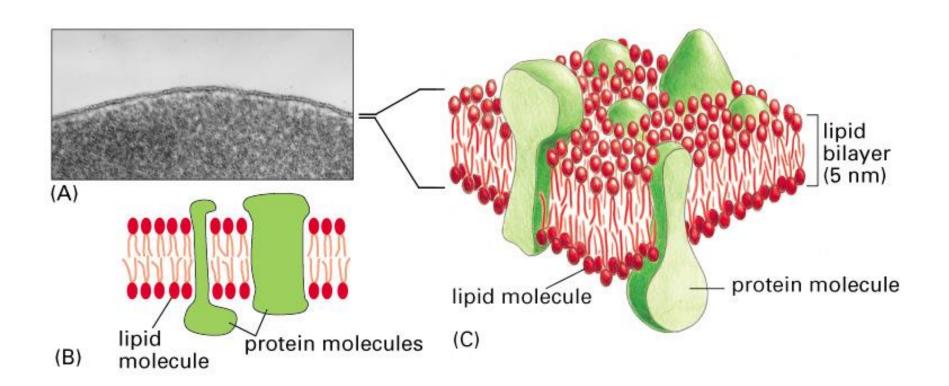
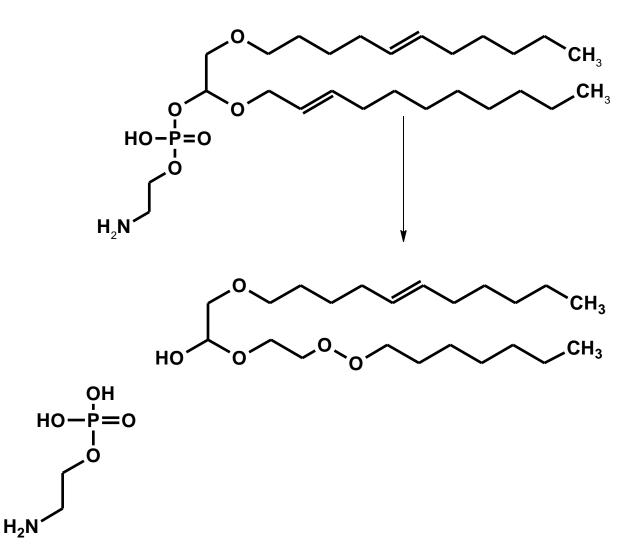


Figure 10–1. Molecular Biology of the Cell, 4th Edition.

Fatty acids chemistry

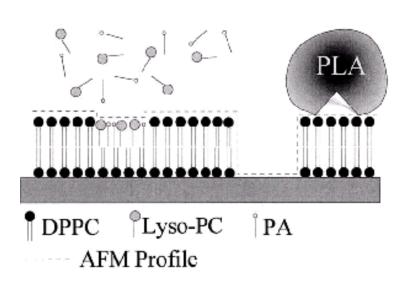
- Hydrolysis of the glycerol backbone
- Oxidation of unsaturated alkyl chains

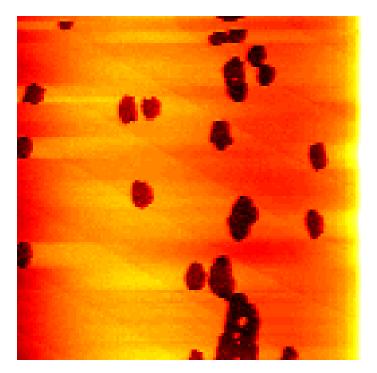




NanoPen







PLA: phospholipase-A

Biomembranes.1420, P.266, (1999)

Cellular chemistry

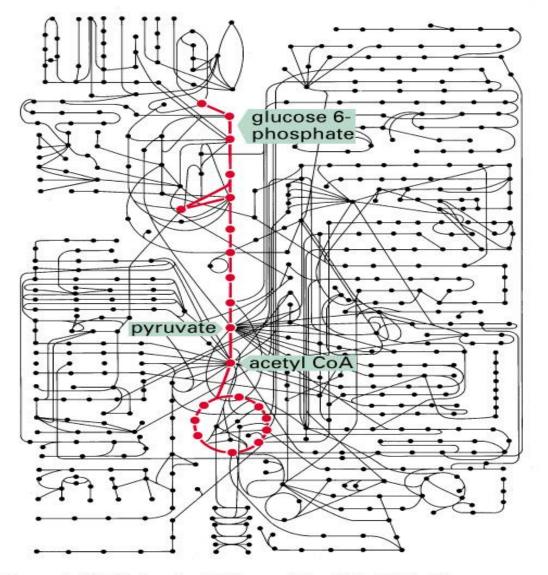




Figure 2–88. Molecular Biology of the Cell, 4th Edition.



The extreme chemistry of life

- pH 0 to10
- NaCl 0 to 2M
- Temp -2 to 110 Celsius
- 0.1 to 110 Mpa
- 0 to 100,000 rem (acceptable dose limit for humans is about 5 rem/year)